	Application No.	Applicant(s)	
Notice of Allowability	10/054.424	WANG ET AL.	
	Examiner	Art Unit	
	Mark Ruthkosky	1745	
The MAILING DATE of this communication. All claims being allowable, PROSECUTION ON THE MERI nerowith (or previously mailed), a Notice of Milowance (PTC NOTICE OF ALLOWABILITY IS NOT A GRANT OF PATE of the Office or upon pelition by the applicant. See 37 CFR.	TS IS (OR REMAINS) CLOSED in DL-85) or other appropriate commits INT RIGHTS. This application is:	n this application. If not included unication will be mailed in due course. TH	
This communication is responsive to 3/12/2004.			
 The allowed claim(s) is/are 1-4 and 7-30. 			
3. The drawings filed on 22 January 2002 are accepted	by the Examiner.		
4. Acknowledgment is made of a claim for foreign prior	rity under 35 U.S.C. § 119(a)-(d)	or (f).	
a) All b) Some* c) None of the:		**	
 Certified copies of the priority documents 			
Certified copies of the priority documents			
Copies of the certified copies of the prior		d in this national stage application from th	
International Bureau (PCT Rule 17.2(a)).			
* Certified copies not received:			
Applicant has THREE MONTHS FROM THE "MAILING D noted below. Failure to timely comply will result in ABANI THIS THREE-MONTH PERIOD IS NOT EXTENDABLE. 5. A SUBSTITUTE OATH OR DECLARATION must be INFORMAL PATENT APPLICATION (PTO-152) with	OONMENT of this application, submitted. Note the attached EX	MINER'S AMENDMENT or NOTICE OF	
B. CORRECTED DRAWINGS (es "replacement sheets"		deciaration is delicient.	
 (a) ☐ including changes required by the Notice of Draf 		v / PTO 048) attached	
1) hereto or 2) to Paper No./Mail Date		V (F10-946) attached	
(b) including changes required by the attached Exar Paper No./Mall Date		in the Office action of	
Identifying indicia such as the application number (see 37 each sheet. Replacement sheet(s) should be labeled as sur	CFR 1.84(c)) should be written on to th in the header according to 37 CF	he drawings in the front (not the back) of R 1.121(d).	
 DEPOSIT OF and/or INFORMATION about the attached Examiner's comment regarding REQUIREN 	deposit of BIOLOGICAL MATE ENT FOR THE DEPOSIT OF BIO	ERIAL must be submitted. Note the DLOGICAL MATERIAL.	
Attachment(s)			
. ☑ Notice of References Cited (PTO-892) . ☐ Notice of Draftperson's Patent Drawing Review (PTO-		formal Patent Application (PTO-152)	
	Pener No.	Interview Summary (PTO-413), Peper No./Mail Date	
□ Information Disclosure Statements (PTO-1449 or PTO Paper No./Mail Date	. –	Amendment/Comment	
. Decaminer's Comment Regarding Requirement for Dep		Statement of Reasons for Allowance	
of Biological Material	9. 🔲 Other	MI Addly 5/20/04	
		Mark Ruthkosky Primary Patent Examiner Art Unit: 1745	

Application/Control Number: 10/054,424 Art Unit: 1745

DETAILED ACTION

Claim Rejections - 35 USC § 103

The rejection of claims 1 and 2 under 35 U.S.C. 103(a) as being unpatentable over Trefilov et al. (PCT/UA94/00018) in view of Gordon et al. (US 5,744,014) has been overcome by the applicant's amendment.

The rejection of claims 3-14 under 35 U.S.C. 103(a) as being unpatentable over Trefilov et al. (PCT/UA94/00018) in view of Gordon et al. (US 5,744,014) as applied to claims 1-2 and further in view of Mototani et al. (US 5,482,798) has been overcome by the applicant's amendment.

Allowable Subject Matter

Claims 1-30 are allowed.

The following is an examiner's statement of reasons for allowance:

Claims 1-15 are to an electrochemical cell capable of producing electrical energy comprising an anode of a zinc anode active material, an aqueous alkaline electrolyte solution comprising potassium hydroxide, a separator and a cathode comprising copper hydroxide, graphitic carbon and a saifur additive selected from sulfur and sulfur compounds. The prior art does not teach an electrochemical cell with a zinc anode, a cathode comprising copper hydroxide, graphitic carbon and a sulfur additive selected from sulfur and sulfur compounds, and an alkaline electrolyte solution comprising potassium hydroxide.

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The most pertinent prior art has been disclosed. Primary and secondary alkaline storage batteries with zinc electrodes are well described in the art as noted in the instant specification. Manganese dioxide is a commonly used cathode active material in these systems. In addition cells with copper hydroxide in the cathode material are known. Trefilov et al. (PCT/UA94/00018), for example, teaches an electrochemical cell comprising an anode active material, an aqueous electrolyte solution, a separator and a cathode comprising copper hydroxide and copper chloride 55-85%, graphitic carbon 5-20% and a sulfur additive 1-25%. Zn is noted as being part of the anode on page 9 and Table 5. The electrolyte is an alkali salt in a water solution (examples.) The reference does not teach the electrolyte to be an alkaline electrolyte of KOH. As noted in the applicant's arguments, this battery is not an alkaline electrolyte battery and one of ordinary skill in the art would not be motivated to change the electrolyte to an alkaline solution of KOH. Light (US 6,207,324 teaches an alkaline cell with a zinc anode and a hydroxide electrolyte, however the cathode is a sulfur electrode without copper hydroxide. The electrolyte does include hydroxides including copper, however, the material is not taught to be the cathode active material. As the prior art does not teach an alkaline battery with a zinc anode. a cathode comprising copper hydroxide, graphitic carbon and a sulfur additive selected from sulfur and sulfur compounds, and an alkaline electrolyte solution comprising potassium hydroxide, the claims are allowed.

Claims 15-30 are to an electrochemical cell comprising an anode active material, an aqueous alkaline electrolyte solution, a separator and a cathode comprising copper hydroxide, carbon nanofibers and a sulfur additive selected from sulfur and sulfur compounds. The prior art does not teach an electrochemical cell comprising an anode active material, an aqueous alkaline electrolyte solution, a separator and a cathode comprising copper hydroxide, carbon nanofibers and a sulfur additive selected from sulfur and sulfur compounds. The most pertinent prior art has been cited. For example, Treflow et al. (PCTTUA9400018) teaches an electrochemical cell comprising a zinc anode active material, an aqueous electrolyte solution, a separator and a cathode comprising copper hydroxide/copper chloride 55-85%, graphitic carbon 5-20% and a sulfur additive 1-25%. The reference does not teach an electrochemical cell comprising a KOH aqueous alkaline electrolyte solution and a cathode comprising copper hydroxide, a sulfur additive selected from sulfur and sulfur compounds and carbon nanofibers as part of the clectrode as a conductive material. Carbon nanofibers are not taught in combination with these materials in the prior art. As such, the claims are allowed.

Any comments considered necessary by applicant must be submitted no later than the payment of the issue fee and, to avoid processing delays, should preferably accompany the issue fee. Such submissions should be clearly labeled "Comments on Statement of Reasons for Allouance"

Examiner Correspondence

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Mark Ruthkorsky whose telephone number is 571-272-1291. The examiner can normally be reached on FLEX schedule (generally, Monday-Thursday from 9.00-6.30.) If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick Ryan can be reached at 571-272-1292. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306. Information regarding the status of an application may be obtained from the Patent

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system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Mark Ruthkosky Primary Patent Examiner Art Unit 1745